



Decolorization of used Dye Baths Applied for Reactive Dyeing of Cotton

Ivana Vojtová^{1,2} and Tomáš Weidlich^{2*}

¹Pleas s.r.o., Havlíčkův Brod, Czech Republic

²University of Pardubice, Faculty of Chemical Technology, Institute of Environmental and Chemical Engineering, Chemical Technology Group, Pardubice, Czech Republic

Abstract

The world's second largest water consumer is the textile industry. Up to 20 % of global industrial water pollution is produced from textile industry [1]. Water is a common liquid medium in textile production, it is used for removing impurities, dyeing, applying finishing agents or for steam production. Only a small amount of used water is vaporized during individual technological operations. The most of applied water in colorization technology is discharged as wastewater. This wastewater typically contains high COD (Chemical Oxygen Demand) and BOD (Biochemical Oxygen Demand) contamination, high amount of total dissolved salts, exhibits high pH and elevated temperature. Wastewater from textile processes is usually highly colored. Effective wastewater treatment with subsequent recycling reduces the amount of consumed water and quantities of discharged pollutants.

This article deals with wastewater recycling based on removal of dissolved organics after reactive dyeing of cotton fabrics. Typically, it is necessary to use the sodium chloride solution to ensure good dye migration into the fiber. Almost all used sodium chloride remains in the wastewater. This article presents research focused on decolorization of wastewater by precipitation of dissolved anionic dyes applying ionic liquid and final treatment by active carbon. This combined decontamination method produces decolorized brine applicable for subsequent dye bath preparation [2].

Keywords: Textile wastewater; Ionic liquid; Reactive dyeing; Cotton

Introduction

Dyeing wastewater exhibits values of pH, COD and amounts of dissolved inorganic compounds, typically sodium chloride. We developed an innovative technique for simple removal of dissolved colorants from this spent dye baths enabling simple recycling of produced brine. Addition of suitable ionic liquid enables effective precipitation of dissolved anionic dyes including hydrolyzed unfixed reactive dyes by formation of insoluble ion pairs. The remaining organic contamination is subsequently removed by applying active carbon. Subsequent removal of insoluble part produces decolorized aqueous NaCl solution.

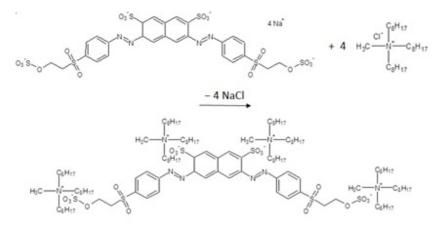


Figure 1: Ion exchange reaction of Novacron Black NN reactive dye and commercially available ionic liquid Aliquat 336 produces insoluble ion pair.



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*Corresponding author: Tomáš Weidlich, University of Pardubice, Faculty of Chemical Technology, Institute of Environmental and Chemical Engineering, Chemical Technology Group, Pardubice, Czech Republic

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Above mentioned ionic liquids are usually defined as organic salts with a melting point below 100 °C. The physical and chemical properties of these liquids are influenced by both the organic cation and the anion. The advantage of using ionic liquids compared with conventional organic solvents is their almost zero vapour tension, low flammability, ability for ion exchange, good solubility for various organic, inorganic and polymeric compounds and high density. Because of these properties, ionic liquids are sometimes referred to as "green solvents" due to their low environmental impact [3] (Figure 1).

ALIQUAT 336 is a commercially available ionic liquid. It is a quaternary ammonium salt used as a phase transfer catalyst and metal extraction reagent, widely used as a phase transfer catalyst for the catalytic oxidation of cyclohexene to 1,6-hexanedioic acid. It finds application in wastewater treatment by using this agent in the treatment of wastewater containing anionic surfactants. It is a colorless viscous liquid with a melting point of -20 °C and a boiling point of 225 °C [4].

Active carbon (Silcarbon) is a material produced from wood or coconut shells. It has a porous structure and a large specific surface area; it has excellent mechanical strength and a high adsorption capacity. Another advantage is its good availability and relatively low cost [5].

Experimental

The dye bath contains around 3wt.% of the reactive dye, 76g/L of sodium chloride, 10g/L of soda ash and 3,2ml/L of caustic soda (50%). The collected wastewater was strongly colored, strongly alkaline (pH=11.8), the sodium chloride content was in the range of 17g/L. To improve the effectiveness of used ionic liquid A336, the pH value was adjusted to a neutral using Brenntaflor AL flocculant (polyaluminium hydroxichloride). After adjusting the pH value to 6.4, 0.3g of ALIQUAT 336 ionic liquid was added to 100mL of spent dye bath and the mixture was stirred for 30 minutes using a magnetic stirrer. After 30 minutes, the solution was filtered producing practically colourless aqueous filtrate containing sodium chloride (Figure 2) (Chart 1&2).



Figure 2: Photo of a decolorized Novacron Black NN dyeing bath; left- wastewater before cleaning, centre- filtrate after treatment with Aliquat 336, right- filtrate after cleaning with Aliquat 336 and Silcarbon.

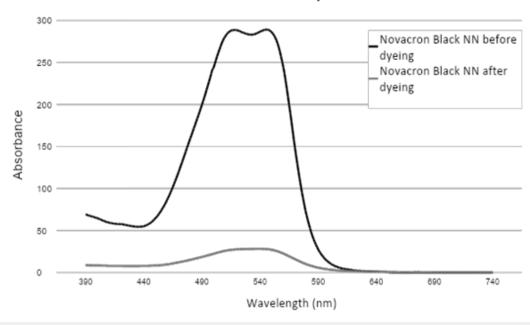


Chart of Novacron Black NN dye exhaustion

Chart 1: VIS spectra of Novacron Black NN dye before and after dyeing.

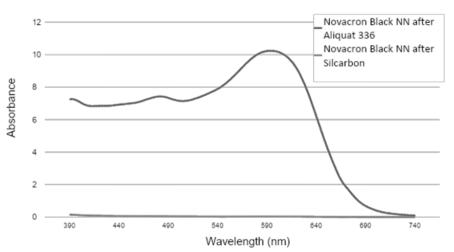


Chart of cleaning of Novacron Black NN dye with Aliquat 336 and with Silcarbon

Chart 2: Removal of Novacron Black NN dye from spent dye bath using Aliquat 336 and Silcarbon.

Conclusion

The non-biodegradable pollution of wastewater after textile dyeing is a major issue in how to improve the impact on the environment. The question remains as to how this can be achieved. The performed experiments support effective decolorization of spent dye baths contaminated with reactive dyes using an ionic liquid with subsequent active carbon treatment.

References

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