



Modeling of Cr³⁺ Doped β-Ga₂O₃ Single Crystals

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Abstract

Crystal field parameters and zero field splitting parameters of Cr^{3+} doped β -Ga₂O₃ single crystals are computed with the help of superposition model. The appropriate sites for Cr^{3+} ions in β -Ga₂O₃ with distortion are adopted for calculation. The experimental values of zero field splitting parameters are in good match with the theoretical ones when local distortion is included into computation. The optical energy band positions for Cr^{3+} in β -Ga₂O₃ are calculated using crystal field parameters and Crystal Field Analysis Program. The results indicate that Cr^{3+} ions substitute at GaII³⁺ octahedral sites in β -Ga₂O₃ single crystals.

Keywords: Superposition model; Crystal field; Zero-field splitting; Optical spectroscopy; $Cr^{3\ast}$ ions in $\beta\text{-}Ga_2O_3$

Introduction

Electron Paramagnetic Resonance (EPR) is a powerful tool for investigating the local site symmetry and Zero Field Splitting (ZFS) of impurity ions in host crystals [1,2]. Superposition Model (SPM) [3-5] is frequently employed to find theoretically the Crystal Field (CF) parameters and Zero Field Splitting (ZFS) parameters.

 β -Ga₂O₃, gallium oxide, is an insulator having a band gap of 4.8eV [6]. When synthesized under reducing conditions, it becomes an n-type semiconductor [7]. Doping yields variation in the conductivity of both the p and n types. These materials have potential applications in optoelectronics These are also used as insulating or conductive window material or substrates [8]. As the above material is stable at high temperatures, it has been widely studied for application as a gas sensor [9]. The electrical conductivity of Ga₂O₃ at large temperatures is changed in the presence of oxidizing or reducing gases. β -Ga₂O₃ has become a good material for optoelectronic devices in the deep ultraviolet region [10]. Gallium oxide crystal doped with trivalent transition ions is of importance in microwave, optical maser and exchange interaction between substitutional ions studies [11]. Introduced impurities giving microscopic structural change affect the optical properties of the crystal based on location of the impurities. The Cr³⁺ion is a very important probe for obtaining information on β -Ga₂O₃ crystal.

The EPR study of Cr^{3+} doped β -Ga₂O₃ has been reported [12]. The EPR spectra of Cr^{3+} ions in β -Ga₂O₃ single crystals were recorded on an X-band spectrometer at room temperature. These spectra in the three mutually perpendicular planes were analyzed employing an effective spin Hamiltonian. The spectroscopic splitting factor g and the Zero Field Splitting (ZFS) parameters of Cr3+ ions were obtained for the octahedral substitutional sites in β -Ga₂O₃ single crystal. The actual local site symmetry of Cr^{3+} ions in the crystal was also investigated.

The laboratory axes (x, y, z) are chosen to be along the crystallographic axes (a, b, c*). The symmetry adopted axes (magnetic axes) are labeled (X, Y, Z). The principal Y axis of g and D tensors of Cr^{3+} ions is found parallel to the crystallographic b axis (monoclinic C2 axis). The principal axes of g and D tensors in the plane normal to the C₂ axis do not coincide. If

an external magnetic field is applied in a plane not containing the C_2 axis, the resonance fields are asymmetric about the extrema [12]. These principal directions and the asymmetry show that the local site symmetry of Cr^{3+} ions in β -Ga₂O₃ crystal is monoclinic. In the principal magnetic axis system, the parameters g, D and E were obtained using EPR-NMR program. There are two chemically distinguishable Ga³⁺ sites in the unit cell: Ga¹³⁺ ions coordinated with four oxygens and Ga23+ ions coordinated with six oxygens [13]. The Cr³⁺ ion EPR spectrum shows a single set of three lines indicating Cr³⁺ ions to be located at only one site. Cr³⁺ ions with an ionic radius 0.0615nm for the six-fold coordination prefer the octahedrally coordinated Ga²³⁺ site (ionic radius 0.062nm) [12].

The present investigation presents the Superposition Model (SPM) of the CF parameters and the ZFS parameters for Cr³⁺ ions in β -Ga₂O₃ crystal. The aim is to obtain the ZFS parameters, the CF parameters and the distortion in the lattice for the Cr³⁺ ions in β -Ga₂O₃ at octahedral sites. The optical energy band positions for Cr³⁺ ions in β -Ga₂O₃ are computed using CF parameters and Crystal Field Analysis (CFA) program. The CF parameters and ZFS parameters evaluated may be useful in future investigations for scientific and industrial applications of such crystals.

Crystal structure

The crystal structure of β -Ga₂O₃ is monoclinic with space group C₂/m [14]. The lattice parameters are a =1.2214nm, b=0.30371nm, c=0.57981nm, β =103.83° and Z=4. Two chemically distinct cation sites are coordinated with oxygens either tetrahedrally or octahedrally. The structure shows double chains of GaO6 octahedra (Ga2) parallel to the b axis, which are connected by GaO4 tetrahedra (Ga1). The crystal structure of β -Ga₂O₃ with symmetry adopted axis system (SAAS) is given in Figure 1.



Figure 1: Crystal structure of β -Ga₂O₃ with symmetry adopted axis system (SAAS).

The Symmetry Adopted Axes (SAA) (local site symmetry axes) are the mutually perpendicular directions of metal-ligand bonds. The Z axis of SAAS is along the metal-ligand bond Ga2-O1 (crystal c*-axis) and the two other axes (X,Y) are perpendicular to the Z axes (in the ab plane) for center I (Figure 1). This shows that Cr^{3+} substitutes for Ga^{23+} in the crystal of β -Ga₂O₃ with approximately orthorhombic symmetry. The ionic radius of Cr^{3+} ion (0.0615nm) [15] is slightly smaller than the ionic radius of Ga^{23+} (0.062nm) suggesting Cr^{3+} ion to sit at the location of Ga^{23+} with certain distortion.

The Cr³⁺ ion position and spherical polar coordinates of the ligands in β -Ga₂O₃ [14] for center I are given in Table 1. These data are used for ZFS and CF calculations for Cr³⁺ ion in β -Ga₂O₃.

Table 1: Fractional coordinates of Cr^{3+} ion (center I) and spherical co-ordinates (R, θ , φ) of ligands in β -Ga₂O₃ single crystal.

De eltiere e Corão	t i ann da	Spherical Coordinates of Ligands				
Position of Cr ² +	Ligands	RA	θ٥	θ٥		
ND: Substitutional (0.15866, 0.50000, 0.31402)	01	1.9373	127.66	-87.30		
	02	2.0742	45.84	-83.27		
	03	1.9358	99.89	0.00		
	01 v	1.9373	127.66	87.30		
	O2 iii	3.3379	130.94	0.00		
	02 v	2.0742	45.84	83.27		
WD: substitutional A (0.49350, 0.55400, 1.15092)	01	6.6197	155.75	22.72		
	02	4.8338	134.82	23.27		
	03	6.9850	137.93	1.54		
	01 v	6.5440	157.28	-18.62		
	O2 iii	6.8470	180.00	4.60		
	02 v	4.7296	136.84	- 19.10		

ND = No distortion, WD = With distortion.

Calculations of zero field splitting parameters

The spin Hamiltonian [16-18] used to find the energy levels of Cr^{3+} ions in crystals is:

$$H = H_{Ze} + H_{ZFS} = \mu_B B.g.S + \sum B_k^q O_k^q = \mu_B B.g.S + \sum f_k b_k^q O_k^q , \quad (1)$$

Where g, μ B and B are the spectroscopic splitting factor, Bohr magneton and static magnetic field, respectively. S represents the effective spin operator and $O_k^q(S_x, S_y, S_z)$ are the Extended Stevens Operators (ESO) [19,20]; B_k^q and b_k^q are the ZFS parameters, fk=1/3 and 1/60 are the scaling factors for k = 2 and 4, respectively. The ZFS terms in (1) for Cr3+ ion (S=3/2) at orthorhombic symmetry sites are given as [21,22]:

$$H_{ZFS} = B_2^0 O_2^0 + B_2^2 O_2^2 = \frac{1}{3} b_2^0 O_2^0 + \frac{1}{3} b_2^2 O_2^2 = D(S_z^2 - \frac{1}{3} S(S+1)) + E(S_x^2 - S_y^2),$$
(2)

The conventional orthorhombic ZFS parameters D, E and $\,B^q_k$, $\,b^q_k$ are related as follows:

$$b_2^0 = D = 3B_2^0, b_2^2 = 3E = 3B_2^2.$$
(3)

The ZFS parameters (in ESO notation) for any symmetry employing SPM [21,22] are obtained as:

$$b_k^q = \sum_i \overline{b}_k(R_0) \left(\frac{R_0}{R_i}\right)^{t_k} K_k^q(\theta_i, \varphi_i), \qquad (4)$$

Where (Ri, θ I, ϕ i) are the spherical polar coordinates of i-th ligand. The intrinsic parameters \overline{b}_k provide the strength of the k-th rank ZFS contribution from a ligand at the distance Ri and the coordination factors K_k^q give the geometrical information. K_k^q for k = 1 to 6 in the ESO notation [23] are presented in Appendix A1 of [24].

Eq. (4) provides conventional ZFS parameters, D and E, in terms of the intrinsic parameters \overline{b}_k , the power-law exponents tk, and the reference distance R0, as [24,25-27]:

$$b_{2}^{0} = D = \frac{\overline{b}_{2}(R_{0})}{2} \left[\left(\frac{R_{0}}{R_{i}} \right)^{t_{2}} \sum_{i} (3\cos^{2}\theta_{i} - 1) \right]$$
(5)

$$b_2^2 = 3E = \frac{b_2^2}{3} = \frac{\overline{b_2}(R_0)}{2} \left[\left(\frac{R_0}{R_i}\right)^{t_2} \sum_i (\sin^2\theta_i \cos 2\varphi_i) \right]$$

Cr³⁺ ion in β-Ga₂O₃ may be supposed to substitute at the Ga²³⁺ ion site, and the interstitial site with similar ligand arrangement. The local symmetry at Cr³⁺ ion site is assumed to be approximately orthorhombic. In octahedral coordination of Cr³⁺ ion in LiNbO3 having Cr³⁺-O2- bond, $\overline{b}_2(R_0)$ =2.34cm⁻¹ and t2=-0.12 [28] were used to obtain b_2^0 and b_2^2 . Because Cr³⁺ ion in β-Ga2O3 has distorted octahedral coordination (Figure 1) with oxygen as ligands, the b_K^q in the present investigation are determined using $\overline{b}_2(R_0)$ =2.34cm⁻¹ 1 and t₂=-1.96 for the center I.

The Cr3+ ion position and spherical coordinates of ligands shown in Table 1 are used for calculation. The conventional ZFS parameters, D and E of Cr^{3+} ion in β -Ga₂O₃ single crystal are evaluated employing Eq. (5). The reference distance R0=0.200nm was taken for the calculation of ZFS parameters [29], and the values are: |D|=88.7×10-4cm-1 and |E|=25.7×10-4 cm-1 for center I. The ratio b_2^2 / b_2^0 should be within the range (0, 1) for orthorhombic symmetry [30]. In the present computation, the ratio $|b_2^2| / |b_2^0|$ =0.870 and |E| / |D| = 0.290 for center I. It is seen that the value of |D| and |E| do not agree with the experimental values though $|b_2^2|/|b_2^0|$ is in the specified range [30]. Hence, with above t2 and reference distance R0, the ZFS parameters |D| and |E| are calculated for Cr³⁺ at the Ga²³⁺ site with distortion having position Ga²³⁺ (0.49350, 0.55400, 1.15092) for center I. The conventional ZFS parameters found now are |D|=5385.2×10-4cm-1, |E|=1288.2×10-4cm⁻¹ for center I, which are in good match with the experimental ones. The ratio $|b_2^2| / |b_2^0| = 0.717$ and |E| / |D| = 0.239 for center I are in the specified range [30]. Further, with above t2 and reference distance R0, the ZFS parameters |D| and |E| are computed for Cr³⁺ at the interstitial site but the values found in this case are inconsistent with the experimental values. Hence, these data are not shown here.

The calculated and experimental ZFS parameters for Cr^{3+} ion in β -Ga₂O₃ are given in Table 2. It is seen from Table 2 that the ZFS parameters |D| and |E| are in good match with the experimental ones [12] when the distortion is introduced into calculation.

Table 2: Calculated and experimental ZFS parameters of Cr3+ doped β -Ga2O3 single crystal for center I along with reference distance.

	D A0	Calculated ZFS Parameters (cm ⁻¹)			Conventional ZFS Parameters (×10 ⁻⁴ cm ⁻¹)		
K ₀ A ^o	b ₂ ⁰	b ² ₂	$ \mathbf{b}_{2}^{2} / \mathbf{b}_{2}^{0} $	D	E	E / D	
ND	2.00	0.00887	0.00772	0.870	88.7	25.7	0.290
Center I WD	2.00	0.53852	0.38648	0.717	5385.2	1288.2	0.23
	·	·	·	<u>`</u>	5385.0°	1288.0 ^e	0.239

ND = No distortion, WD = With distortion, ^e = experimental.

Calculations of crystal field parameters

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The CF energy levels of transition ions in crystals [31-34] using Wybourne operators [15,16,35] are given by:

$$H_{CF} = \sum_{k} B_{kq} C_q^{(k)} \qquad (6)$$

Where \mathcal{H} CF is CF Hamiltonian. The CF parameters in (6) for a metal-ligand complex are determined using SPM [21,22] as:

$$B_{kq} = \sum_{i} \overline{A}_{k} \left(\frac{R_{0}}{R_{i}} \right)^{l_{k}} K_{kq}(\theta_{i}, \varphi_{i}) .$$
 (7)

Where R0 is the reference distance; R_{μ} , θ_{μ} , φ_{i} are the spherical polar coordinates of the ith ligand and Kkq are the coordination factors [31]. To obtain B_{kq} (k=2, 4; q=0, 2, 4); \overline{A}_{2} = 40, 400cm⁻¹, t_{2} = 1.3, \overline{A}_{4} = 11, 700cm⁻¹ and t_{4} = 3.4 are used [31]. The calculated B_{kq} parameters are presented in Table 3. The ratio B_{2}^{2}/B_{2}^{0} = 0.255 for

center I, which indicates that Bkq parameters are standardized [30]. Taking B_{kq} parameters given in Table 3 and CFA program [32,33], the CF energy levels of Cr^{3+} ion in β -Ga₂O₃ single crystals are calculated by diagonalizing the complete Hamiltonian. The calculated energy values are given in Table 4. The calculated energy

values are compared with the experimental energy values for Cr³⁺: β -Ga₂O₃ [36]. From Table 4, it is noted that the theoretical and experimental band positions are in reasonable match. Thus the theoretical study of Cr³⁺ ions at Ga²³⁺ sites in β -Ga₂O₃ supports the experimental one [12,36].

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Table 3: B_{k0} parameters of Cr^{3+} doped β -Ga₂O₃ single crystal for center I with distortion.

Calculated Bkq (cm ⁻¹) Parameters Used for CFA package							
	R ₀ A°	B ₂₀	B ₂₂	B ₄₀	B ₄₂	B ₄₄	B_{22}/B_{20}
Center I WD	2.00	43690.67	11166.53	-124.712	2332.583	411.8434	0.255

WD = With distortion.

Table 4: Experimental and calculated energy band positions (centers I) of Cr^{3+} doped β -Ga₂O₃ single crystal.

Transition from ⁴ A _{2g} (F)	Experimentally Observed band (cm ⁻¹⁾	Calculated Energy Band from CFA (cm ⁻¹) Center I		
² E _g (G)		7464,12004		
${}^{2}\mathrm{T}_{1\mathrm{g}}\left(\mathrm{G}\right)$	16807	12097, 13723, 14448		
${}^{4}\mathrm{T}_{2\mathrm{g}}\left(\mathrm{F} ight)$		17233, 17380, 17578, 17731, 19213, 19347		
⁴ T _{1g} (F)	23923	23047, 23206, 23725, 26704, 27267, 31456		
${}^{2}\mathrm{T}_{1\mathrm{g}}(\mathrm{aD})$		34590, 35361, 37334		
² E _g (bD)		45556, 45716		

(Racah parameters A, B and C, spin-orbit coupling constant and Trees correction are 0, 726.3, 2905.2 (= 4B), 276 and 70cm⁻¹, respectively)

The optical absorption/excitation spectra of Cr^{3+} -activated phosphors have been explained with the help of Franck-Condon analysis with the Configurational-Coordinate (CC) model [37]. The different excited state-ground state transitions in Cr^{3+} are because of the strong coupling with the lattice vibrations (CC model) [37]. The CC model has not been employed and hence there is difference between excited-state peak energies found here and Zero-Phonon Line (ZPL) energies discussed in [37,38]. The Cr3+-activated oxide phosphors are classified into two groups: O-Cr-A and O-Cr-B [38]. Phosphors of O-Cr-A type have an energy inequality relation of E(2Eg)ZPL < E(4T2g)ZPL, giving a series of the sharp emission peaks by the $2Eg \rightarrow 4A2g$ transitions. Phosphors of O-Cr-B type have an energy inequality relation of E(4T2g) ZPL < E(2Eg) ZPL, giving a broad emission band due to the $4T2g \rightarrow 4A2g$ transitions. β -Ga₂O₃: Cr³⁺ comes under O-Cr-B type phosphors [38].

Summary and Conclusion

The Zero-Field Splitting (ZFS) parameters and Crystal Field (CF) parameters for Cr³⁺ ions in β -Ga₂O₃ single crystals are calculated with the help of Superposition Model (SPM). Cr³⁺ ions in β -Ga₂O₃ crystal at Ga²³⁺ ion sites, interstitial site and distortion models are considered for calculation. The calculated conventional ZFS parameters for Cr³⁺ ion at Ga²³⁺ sites in β -Ga₂O₃ single crystal provide good agreement with the experimental values when distortion is included in the calculation. It is noted that the Cr³⁺ ions substitute at Ga²³⁺ ion sites in β -Ga₂O₃ lattice. The CF energy values for Cr³⁺ ions at Ga²³⁺ sites computed using CFA package and CF parameters are in reasonable match with the experimental study.

Modeling procedure employed in this investigation may be useful in future to correlate EPR and optical data for different ion-host systems in exploring crystals for various scientific and industrial applications.

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