

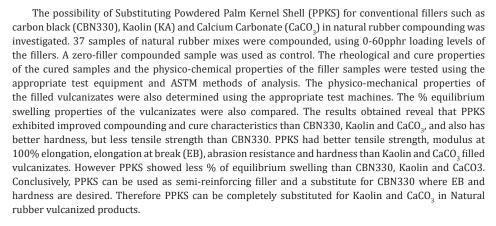
# Substitutional Studies of Powdered Palm Kernel Shell (PPKS) for Carbon Black, Kaolin and Calcium Carbonate as Fillers in Natural Rubber Compounding

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#### **Abstract**



**Keywords:** Substitution; Fillers; Powdered palm kernel shell; Carbon black; Physico-chemical; Physico-chanical

### Introduction

Natural rubber crumbs have always been the major input in rubber compounding, while reinforcing fillers such as carbon black, (CaCO<sub>3</sub>) and Kaolin are the conventional non-renewable fillers used. Suitable replaceable fillers that are environmentally friendly, non-health hazard causing, and do not cause environmental pollution, that can replace the non-renewable conventional filler, such as CBN330 are constantly being researched and sourced for [1]. The utilization of cocoa-pod husks and rubber seed shells as fillers in natural rubber compounding have been studied [2]. Equally, the mechanical properties of natural rubber reinforced by agricultural waste products have been evaluated [3]. Rice husks have been discovered as potential filler in epoxidized natural rubber [4]. Calcium Carbonate (CaCO<sub>3</sub>) has been studied as filler with flame retardant properties in Styrene-butadiene rubber [5]. Carbonized plantain peel has been investigated to substitute N330 Carbon black, to improve the cure characteristics, physico-mechanical and equilibrium swelling properties in natural rubber vulcanizates [6]. Rheological rubber compounded with blend of carbonized melon seed shell and carbon black has also been investigated [7]. Tensile strength properties of oil palm empty fruit Bunch-Polyurethane composite have been Studied [8].

From all of the above findings, there has not been any suitable replacement or substitution for carbon black, especially in the area of improved tensile strength in natural rubber





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compounding applications. This has prompted the investigation of PPKS, a low cost agricultural waste renewable material that is environmentally friendly, as suitable filler for replacement of conventional filler (like CBN330, Kaolin and  $CaCO_3$ ) in natural rubber compounding.

#### **Materials and Methods**

#### **Materials**

Materials used and their sources are as follows: Natural rubber crumb (NSR-10) was obtained from Omatseye Rubber Factory, Sapele, Delta State; Stearic acid; Zinc Oxide; processing oil (Flectol), Sulphur, TMTD, were obtained from British Drug House, Poole, England; Carbon Black (N330), was obtained from NNPC Carbon Black Plant, Ekpan-Warri, Delta State, Nigeria; Calcium Carbonate and Kaolin were obtained from Salami Orogho Quarry Company,

Okpilla, Edo State, Nigeria; Palm Kernel Shell, obtaind from Palm Estate, Mosogar, Sapele, Delta State, Nigeria.

#### Methods

**Material preparations:** The Palm Kernel shell (PKS) were ground into fine particle sizes of 60, 106 and 230um respectively, using a piston and mortar and a screen sieve of appropriate particle sizes. They were stored for further use. Kaolin and CaCO<sub>3</sub> were also screened to the same particle size as the PKS. The N330 Carbon Black was used as obtained from NNPC, Warri and Delta State.

Preparation of the natural rubber mixes (compounding): The formulation of 37 mixes in gms were prepared, varying the

loadings of the fillers and particle sizes, while the zero filler loading was used as a control sample as shown in Table 1. The best fillers of 30g loading samples mixes were used for further test analysis.

**Table 1:** Natural rubber/filler formulation mixes.

Material	Filler						
	Zero filler	CBN330	PPKS	CaCO <sub>3</sub>	PPKS		
N Rubber	100	100	100	100	100		
Stearic acid	2	2	2	2	2		
Zinc oxide	6	6	6	6	6		
Electo H (TMQ)	2	2	2	2	2		
MBT	1	1	1	1	1		
Processing oil (Flextol)	10	10	10	10	10		
Sulphur	2	2	2	2	2		
Carbon black	-	30	-	-	-		
Kaolin	-	-	-	-	30		
CaCO <sub>3</sub>	-	-	-	30	-		
PPKS	-	-	30	-	-		
Mixing time (mins)	10	17	18	19	22		

## **Experimental Test Procedure**

The physico-chemical analysis of the various fillers in terms of the Percentage Ash and moisture content, iodine adsorption number, bulk density, PH value were done using the conventional methods (ASTM) [9].

The cure characteristics, that is thee rheological (flow) properties analysis were done using the oscillating disc rheometer. The physico-mechanical analysis which include Hardness, Abrasion resistance, Tensile Strength, 100% Elongation, and Percentage equilibrium swelling tests were done, using the appropriate test machines and standard procedure of ASTM [10].

## **Results and Discussions**

The physico-chemical properties of the fillers for different materials are shown in Table 2, while the cure characteristics result is shown in Table 3. From the result as shown in Table 2, PPKS has higher acidity, moisture content and bulk density, but less loss on ignition and iodine adsorption number than CB (N330). The rheological properties of PPKS filled vulcanizate improved over CB

(N330) which accommodated the strain of the polymer mix with the size and loading of the filler [11].

**Table 2:** Physico-chemical properties of the fillers.

Danamatan	Filler						
Parameter	PPKS	CB (N330)	Kaolin	CaCO <sub>3</sub>			
Loss of ignition at (350 °C)	65.58	71.0	59.0	53.0			
Moisture content at (1,280 °C)	13.6	5.32	8.15	18.54			
Ph Value	6.80	6.50	8.60	8.50			
Iodine adsorption No. (ml/g)	4.40	4.71	4.07	4.30			
Mesh size (um)	60.0	35.0	60.0	60.0			
Bulk density (gm/v)	0.54	0.30	0.75	0.90			

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Table 3: Rheological (cure) properties.

Filler Loading (pphr)	Scorch Time (mins)			Cure Time (mins)				
	PPKS	CB N330	Kaolin	CaCO <sub>3</sub>	PPKS	CB N330	Kaolin	CaCO <sub>3</sub>
0	2.50	2.40	4.11	4.20	5.14	5.48	6.19	6.20
10	2.07	2.37	3.30	2.35	5.01	6.13	6.46	4.58
20	2.17	3.09	3.29	3.14	5.32	6.28	8.05	5.50
30	2.10	2.44	4.30	3.01	4.32	6.05	10.18	5.40
40	2.34	2.41	5.59	2.58	5.07	5.46	10.54	5.40
50	2.42	2.42	5.48	2.31	5.48	5.50	10.69	5.08
60	2.55	2.11	3.53	2.52	6.08	5.50	10.75	5.36

Better rheological properties were also obtained for PPKS filled vulcanizates over  $CaCO_3$  and Kaolin, as shown in Table 3. The physico-mechanical properties of PPKS filled vulcanizates were compared against CB (N330),  $CaCO_3$  and Kaolin as shown in Table 4.

The results obtained show comparable tensile strength of PPKS and CB (N330) fillers, while PPKS fillers exhibited better modulus at 100% elongation, EB, hardness, abrasion resistance and percentage equilibrium swelling properties over CB (N330), CaCO<sub>2</sub> and Kaolin.

Table 4: Physico-mechanical properties.

Downwater	Filler						
Parameter	Zero filler	PPKS	CB(N330)	Kaolin	CaCO <sub>3</sub>		
Tensile strength (Mpa)	17	24	27	22	21		
100% Elongation	10	11	13	11	10		
Elongation at break (%)	285	330	285	315	260		
Hardness (IRHD)	39	54	47	36	36		
Abrasion resistance (mg/1000)	0.54	0.18	0.17	0.89	0.14		

Improved tensile strength was observed in PPKS filler over  $CaCO_3$  and Kaolin. The results obtained may be explained that fillers with higher carbon content, provide greater reinforcement than those with lower carbon content, because carbon itself is a good reinforcing filler [11], which is exhibited in better reinforcing ability of CB (N330) over PPKS filler.

The tensile modulus and hardness are controlled by the surface activity and polymer mix-filler interaction as reported by [11]. These properties have been exhibited by the PPKS vulcanized filler over CB (N330),  ${\rm CaCO}_3$  and Kaolin as can be observed in the results obtained in Table 4.

**Table 5:** % equilibrium swelling properties.

Filled	Solvent				
Vulcanizate	Kerosene	Benzene	Diesel		
Zero filler	40	95	40		
CN(N330)	35	48	36		
Kaolin (KA)	60	78	45		
CaCO3	40	85	40		
PPKS	30	45	30		

The results of the low % equilibrium swelling in PPKS filled NR vulcanizate as shown in Table 5, which is consistent with earlier studies [12], that low equilibrium swelling may be due to filler related parameters, such as particle size, acidity and moisture content, which are critical factors for forming interfacial bond with NR Matrix. These properties were exhibited by PPKS filled

vulcanizates, hence the improved performance over CB (N330), Kaolin and CaCO<sub>3</sub> fillers vulcanizates, were observed. The least % equilibrium were obtained in Kerosene and highest in benzene due to the smaller size benzene molecules having higher diffusion rate than individual molecule of kerosene and diesel which are mixtures of alkane hydrocarbon [13,14].

## Conclusion

In conclusion, PPKS filler has been found to be semi-reinforcing filler with slightly less tensile properties than CB (N330). Subsequently, PPKS filler can be substituted for CB (N330) where the performance properties such as elongation at break, hardness and abrasion resistance are desired. PPKS filler also exhibited better % equilibrium swelling properties over CB (N330), Kaolin and  $\rm CaCO_3$  and can be used as good swelling resistance vulcanized NR product of benzene, kerosene and diesel solvents. Finally, PPKS filler can be conveniently used as a substitute for Kaolin and  $\rm CaCO_3$  fillers in NR compounding to produce relevant rubber moldable products.

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