4,4-Dimethoxy-2-butanone as 1,3-Dielectrophilic 3-Carbon Building Block: New Route for the Synthesis of Toluene, o-Xylene, Naphthalenes and Pyrimidines

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Abstract

4,4-dimethoxy-2-butanone 1 is shown to react with allyl magnesium chloride/allyl bromide in the presence of zinc to yield the corresponding toluene in 63% and 52% yields respectively. Similarly, crotyl bromide reacted with 1 in the presence of zinc to yield o-xylene in 32% yield. 1 was also reacted with benzyl magnesium chloride and phenyl acetonitrile to yield the corresponding naphthalenes in 53% and 61% yields respectively. The ketone 1 was reacted with guanidines to yield the corresponding pyrimidines in 58% and 66% yields respectively.

Keywords: 4,4-dimethoxy-2-butanone; Zinc; Toluene; Naphthalene; Pyrimidine

Introduction

The 4, 4-Dimethoxy-2-butanone 1 (Figure 1) is an interesting three carbon 1, 3-dielectrophilic building block. Which is commercially available and a few of its synthetic applications have been reported in recent years. One of the earliest reactions [1] of 1 is reported to have reacted with hydrazine hydrate to yield the corresponding 3-methylpyrazole in excellent yield. Recently there are a number of patents [2-5] described on the reaction of various phenyl hydrazines with 1 to obtain the corresponding aryl methyl pyrazoles in around 60-68% overall yields. In one of their publications Nielsen et al. [6] have reacted p-methyl benzyl magnesium chloride with 1 to give initially the corresponding carbinol acetal in good yields which on treatment with hydrobromic acid in acetic acid, the corresponding 3, 7-dimethyl naphthalene was formed in 24% yield. Similarly, Butterfield and coworkers [7] as a part of their total synthesis of corannulene reacted m-methyl benzyl magnesium chloride with 1, initially to afford the corresponding carbinol which on treatment with acetic acid and 0.8M sulfuric acid at room temperature the corresponding 2, 7-dimethyl naphthalene was formed in 71% yield. Since many years we have been working on α-oxoketenedithioacetal chemistry [8-18] as a useful class of 1, 3-dielectrophilic building blocks for the synthesis of a variety of 5 and 6 membered heterocycles and aromatics. We wanted to extend these ideas to use 4, 4-dimethoxy butanone as a useful 1, 3-dielectrophilic building block and some of these selected reactions are reported in this communication.

Results and Discussion

To begin with the allyl magnesium bromide [19] 2 was reacted with 1 at room temperature to afford the corresponding carbinol acetal 4 in 90% yields. Similarly, crotyl bromide reacted with 1 in the presence of zinc 53% yield (Scheme 1). Thus, allyl bromide 3 was reacted with 1 in the presence of zinc [20] and saturated ammonium chloride when the corresponding toluene was obtained in 70% yield. The acetal was then refluxed with boron trifluoride etherate as described above to afford the expected toluene 5 (Scheme 1) in 52% yield. The structure of toluene was fully confirmed by its super imposable IR spectra and other analytical data.
Scheme 1: Synthesis of toluene from method 1 and method 2.

We next extended this reaction for the synthesis of p-xylene (Scheme 2) by reacting crotyl zinc bromide \([21]\) and it was reasoned that the organo zinc bromide prepared from crotyl bromide would react with 1b in 1,2 addition mode to yield the corresponding carbinol acetal 8, which on acid assisted cyclization would give p-xylene 9 (path a). However, the anion reacted from its third carbon with 1b to yield the different carbinol acetal 10 which on cyclization the corresponding o-xylene 11 (path b) was obtained in 32% yield (Scheme 2).

Scheme 2: Synthesis of o-xylene.

In the next experiment, benzyl magnesium chloride \([22]\) 12 was reacted with 1 to afford the corresponding carbinol acetal 13 in high yield. It was then cyclized in the presence of boron trifluoride etherate, the corresponding 2-methyl naphthalene 14 (Scheme 3) was obtained in 53% yield.

Scheme 3: Synthesis of 2-Methylnaphthalene.

Also, phenyl acetonitrile \([23]\) 15 was reacted with 1 in the presence of sodium hydride (Scheme 4) using a mixture of dimethylformamide and tetrahydrofuran (1:4) to afford the corresponding 1,4-addition-elimination product 16 which on refluxing with boron trifluoride etherate, the corresponding 1-methyl-4-cyanonaphthalene 17 was obtained in 61% yield (Scheme 4).

Scheme 4: Synthesis of 1-Methyl-4-cyanonaphthalene.

The building block 1 was also reacted with guanidine nitrate \([24-28]\) in the presence of sodium ethoxide in boiling ethanol to give the corresponding 2-amino-4-methylpyrimidine 18 in 66% yield (Scheme 5).

Scheme 5: Synthesis of 2-Amino-4-methylpyrimidine and 2-Benzylamino-o-4-methylpyrimidine.

Similarly, when benzyl guanidine \([24-28]\) was reacted with 1 as described above to afford the corresponding 2-benzylamino-o-4-methylpyrimidine 19 in 58% yield (Scheme 5).

Experimental

General instructions

All the reagents and solvents used were purchased from commercial suppliers (Aldrich, TCI and SD fine chemicals) and used without further purification. 1H and 13C NMR spectra were recorded on 400MHz and 500MHz spectrophotometer using CDCl3 as the solvent. Chemical shift values are reported in ppm taking tetramethylsilane as the internal standard, and J values are given in hertz. HRMS data were acquired on a time of flight (TOF) mass spectrometer (Waters Q-TOF Premier-HAB213). IR spectra of all the compounds were obtained by PerkinElmer Spectrum version 10.3.06 IR spectrometer. Reactions were monitored by thin layer chromatography (TLC) using precoated silica gel plates, visualized by UV light. Melting points were determined using Thiele tube apparatus and are uncorrected.

Experimental Procedure

Procedure for the synthesis of Toluene 5.

Method 1: Synthesis of Toluene from allyl magnesium bromide: In a neat reaction under
argon, allyl magnesium bromide 2 (22.7 ml) was slowly added to 1 (3 g, 22.7 mmol) at 0 °C and the reaction mixture was stirred at that temperature for 45 min. (TLC). The reaction mixture was treated with saturated ammonium chloride solution (10 ml) followed by extraction with ether which was dried, and the solvent was removed under reduced pressure to afford the corresponding carbinol 4 (3.53g, 20.28mmol) in 90% yield. The carbinol acetal was then refluxed with boron trifluoride etherate (6.18g, 50.65mmol) for 30 minutes (TLC). The reaction mixture was poured into saturated NaHCO₃ solution and extracted with ether (30ml), washed (H₂O), dried (Na₂SO₄) and evaporated to yield the crude toluene 5 which was purified through vacuum distillation to give pure toluene in 63% yield; [Found: ^1H NMR (400 MHz CDC13) 2.34 (3H, s, CH₃), 7.25-7.14 (5H, m, ArH); ^13C NMR (99.5 MHz, CDC13) 21.5, 125.4, 128.3, 129.1, 138.0; HRMS: MH⁺, found 93.0703. C, H, requires 93.0704.

Method 2: Synthesis of Toluene from allyl bromide: To the mixture of butanone 1 (4g, 30mmol), crotyl bromide (8g, 60mmol), Zn powder (3.92g, 60mmol) and THF (3ml), saturated NH₄Cl solution (4ml) was added. The reaction mixture underwent exothermic reaction for 20 minutes to give carbinol 2 (2.76g, 15.9mmol) in 70% yield. The carbinol acetal was then refluxed with boron trifluoride etherate (4.83g, 39. mmol) for 30 minutes. The reaction mixture was poured into saturated NaHCO₃ solution and extracted with ether (30ml), washed (H₂O), dried (Na₂SO₄) and evaporated to yield the crude toluene 5 which was purified through vacuum distillation to afford the desired toluene 2 in 52% yield. The structure was confirmed by superimposable IR spectra. [Found: ^1H NMR (400 MHz CDC13) 2.34 (3H, s, CH₃), 7.25-7.14 (5H, m, ArH); ^13C NMR (99.5 MHz, CDC13) 21.5, 125.4, 128.3, 129.1, 138.0; HRMS: MH⁺, found 93.0703. C, H, requires 93.0704.

Procedure for the synthesis of o-Xylene 11: To the mixture of butanone 1 (4g, 30mmol), crotyl bromide (8g, 60mmol), Zn powder (3.92g, 60mmol) and THF (3ml), saturated NH₄Cl solution (4ml) was added. The reaction mixture underwent exothermic reaction for 20 minutes to give carbinol 10 (2.84g, 15mmol) in 48% yield. The carbinol acetal was then refluxed with boron trifluoride etherate (5.36g, 37.7mmol) for 30 minutes. The reaction mixture was poured into saturated NaHCO₃ solution and extracted with ether (30ml), washed (H₂O), dried (Na₂SO₄) and evaporated to yield the crude o-xylene which was purified through controlled vacuum distillation to afford the desired o-xylene 11 in 32% yield. The structure was confirmed by superimposable IR spectra. [Found: ^1H NMR (400 MHz CDC13) 2.34 (3H, s, CH₃), 7.25-7.14 (5H, m, ArH); ^13C NMR (99.5 MHz, CDC13) 19.8, 125.9, 129.7, 131.7, 133.7, 135.4.]

Procedure for the synthesis of 2-Methyl-4-cyanonaphthalene 17: To a well stirred suspension of sodium hydride (0.60g, 25.25mmol, 60% purity) in DMF (5ml) and toluene (10ml) at 0 °C, a solution of benzyl cyanide 15 (1.15g, 9.84mmol) in toluene (5ml) was added drop wise and continued stirring for 15 minutes. The butanone 1 (1g, 7.57mmol) in toluene (5ml) was then added slowly and the reaction mixture was stirred at room temperature for 6h (TLC). It was then poured into saturated ammonium chloride solution and extracted with ethyl acetate (30X2). The combined organic extracts was washed with water, dried (Na₂SO₄) and evaporated to yield crude 1, 4-addition elimination adduct 16. This adduct was treated with boron trifluoride etherate (2.08g, 17mmol) and refluxed for 30 min (TLC). The reaction mixture was then poured into saturated solution of sodium bicarbonate (20ml) and extracted with ethyl acetate, washed (H₂O) dried (Na₂SO₄) and the solvent was removed to give the corresponding 2-methyl-4-cyanonaphthalene 17 which was passed over silica gel column eluting with petroleum ether and ethyl acetate (96:4) to afford colorless liquid in 61% yield; [Found: IR (solid film) 3406, 3041, 2927, 2219, 1719, 1163, 758cm⁻¹; ^1H NMR (400 MHz CDC13) 2.65 (3H, s, SCH₃), 7.52 (1H, t, J=8.0 Hz, ArH), 7.98 (1H, d, J=8.0 Hz, ArH), 8.13 (1H, d, J=8.4 Hz, ArH); ^13C NMR (125.76 MHz, CDCl₃) 16.1, 45.6, 106.1, 131.7, 133.7, 135.4.]

Procedure for the synthesis of 2-Amino-4-methylpyrimidine 18: A mixture of sodium ethoxide (2.06g, 30.3mmol) and guanidine nitrate (1.38g, 11.3mmol) in ethanol was stirred for 20 minutes. Then a solution of butanone 1 (1g, 7.57mmol) was added and the reaction mixture was refluxed for 6h (TLC). The solvent was removed under reduced pressure and the reaction mixture was treated with water and extracted with ethyl acetate, washed (H₂O), dried (Na₂SO₄) and the solvent was removed under vacuum to get the crude pyrimidine 18 which was passed over silica gel column eluting with petroleum ether and ethyl acetate (94:6) to yield the colorless crystalline needles in 66% yield. [Found: mp 150-152 °C; IR (Liquid film) 3297, 3154, 3001, 2154, 1563,1211cm⁻¹ ; ^1H NMR (400 MHz CDC13) 2.33 (3H, s, CH₃), 5.01 (2H, s, NH₂), 6.48 (1H, d, J=5.2 Hz, ArH), 8.13 (1H, d, J=4.8Hz, ArH); ^13C NMR (99.5 MHz, CDCl₃) 24.1, 111.4, 158.0, 163.0, 168.5; HRMS: MH⁺, found 110.0717. C₅H₈N₃ requires 110.0718.

Procedure for the synthesis of 2-Benzylamino-4-methylpyrimidine 19: To a well stirred suspension of sodium ethoxide (2.06g, 30.3mmol) in ethanol and benzyl guanidine sulfate (2.8g, 11.36mmol) at room temperature for 20minutes. A solution of butanone 1 (1g, 75.7mmol) in ethanol was added and the reaction mixture was refluxed for 6h (TLC). The reaction mixture after work up with cold water and extraction with ethyl acetate, washed (H$_2$O), dried (Na$_2$SO$_4$) and evaporated to yield the corresponding crude pyrimidine 19 which was passed through silica gel column and eluting with petroleum ether and ether (94:6) to yield the colorless crystalline needles in 58% yield. [Found: mp 70-72 °C; IR (KBr pellet) 3271, 3030, 2920, 1948, 1567 cm$^{-1}$; $^1$H NMR (400 MHz CDCl$_3$) 2.31 (3H, s, CH$_3$), 4.62 (2H, d, J=5.96 Hz, CH$_2$), 5.48 (1H, s, NH), 6.41 (1H, d, J=5.04 Hz, ArH), 7.35-7.22 (5H, m, ArH), 8.10 (1H, d, J=4.56 Hz, ArH); $^{13}$C NMR (199.5 MHz CDCl$_3$) 24.2, 45.4, 45.5, 110.6, 127.2, 127.6, 128.6, 139.4, 157.7, 162.3, 168.1; HRMS: MH$^+$, found 200.1114. C$_{12}$H$_{14}$N$_3$ requires 200.1188.

Conclusion

The 4, 4-Dimethoxy-2-butanone is commercially available and consequently the chemistry based on its reactivity has been reported sparingly and the present communication explains some useful important reactions for the synthesis of aromatic and heteroaromatic compounds.

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